

**Hygrophiloside tetra-acetate (1a).** Prepared by acetylation of 1 ( $\text{Ac}_2\text{O}$ , pyridine, room temp., 2 hr). Crystallization from EtOH gave pure 1a, mp 148–148.5°;  $[\alpha]_D^{20} - 69^\circ$  ( $\text{CHCl}_3$ ;  $c$  0.9);  $^1\text{H NMR}$  (90 MHz,  $\text{CDCl}_3$ ):  $\delta$  9.60 (s, CHO-10), 6.78 (m, H-7), 6.23 (d,  $J = 2$  Hz, H-1), 6.12 (d,  $J = 6.5$  Hz, H-3), 4.92 (d,  $J = 6.5$  Hz, H-4), 3.39 (m, H-9), 3.12 (s, OH), 2.91 and 2.68 (br AB-system,  $J = 19$  Hz,  $\text{CH}_2$ -6), 2.11, 2.03, 2.00 and 1.98 (4  $\times$  OAc);  $^{13}\text{C NMR}$  (22.6 MHz,  $\text{CDCl}_3$ ):  $\delta$  189.3 (C-10), 152.3 (C-7), 142.9 (C-8), 139.4 (C-3), 110.0 (C-4), 91.4 (C-1), 73.1 (C-5), 52.7 (C-9), 46.0 (C-6), 96.0, 71.0, 71.8, 68.2, 71.8, 61.5 (C-1–C-6 in the  $\beta$ -glucopyranosyl moiety). (Found: C, 52.7; H, 5.5.  $\text{C}_{23}\text{H}_{28}\text{O}_{13} \cdot 1/2 \text{H}_2\text{O}$  requires: C, 53.0; H, 5.6%.)

**Conversion to isoaucubin penta-acetate (2a).** To a stirred soln of 1 (90 mg) in MeOH (5 ml) was added  $\text{NaBH}_4$  (60 mg). After 45 min the mixture was taken to dryness and the residue acetylated as above. Work-up gave, after prep. TLC, 2a (78 mg) as crystals from EtOH, mp and mmp 121–122°;  $[\alpha]_D^{20} - 91^\circ$  ( $\text{CHCl}_3$ ;  $c$  0.7), lit. [5] mp 125°;  $[\alpha]_D^{28} - 46^\circ$  (EtOH;  $c$  0.95); an authentic sample, kindly supplied by Dr. Endo, exhibited the

rotation  $[\alpha]_D^{20} - 90^\circ$  ( $\text{CHCl}_3$ ;  $c$  0.2).

The  $^1\text{H}$  and  $^{13}\text{C NMR}$  spectra were as reported [5].

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# A NOR-SESQUITERPENE- $\gamma$ -LACTONE FOUND IN *CREPIS PYGMAEA*

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**Key Word Index**—*Crepis pygmaea*; Compositae; sesquiterpenoid lactones; 1,2,4,5-tetrahydro-11-nor-11-hydroxy- $\Delta^{7,11}$ -santonin.

**Abstract**—The isolation and structural elucidation of a novel nor-sesquiterpene- $\gamma$ -lactone are reported.

## INTRODUCTION

We have recently reported the isolation and structural elucidation of an unusual nor-sesquiterpene- $\gamma$ -lactone (1) found in *Crepis pygmaea* L.‡ [1], whose structure has been definitively confirmed through synthesis [2]. During the isolation of 1 from the chloroform extract of the whole plant, a number of minor by-products were observed.

The present report describes the characterization of one of these products as the novel nor-sesquiterpenoid, 1,2,4,5-tetrahydro-11-nor-11-hydroxy- $\Delta^{7,11}$ -santonin (2) (or its mirror image).

## RESULTS AND DISCUSSION

Compound 2 was obtained by chromatographic fractionation of the chloroform extract of *C. pygmaea* as previously described [1]. In particular, repeated fractionation on silica gel (Kieselgel 60 Merck) columns of the crude fraction containing 1 [1] yielded, besides 1, chromatographically pure 2 (80 mg), which crystallized from ether as white cubes.

Compound 2, mp 177–179°;  $R_f$  (TLC) 0.39 (pre-coated Merck plates,  $n$ -hexane- $\text{CH}_2\text{Cl}_2$ - $i$ -PrOH, 34:6:10),  $[\alpha]_D^{20} + 25.7^\circ$  ( $\text{CHCl}_3$ ;  $c$  2.9); MS  $m/z$  250  $[\text{M}]^+$ ; UV  $\lambda_{\text{max}}^{\text{EtOH}}$  nm (s): 237 (8420); molecular formula  $\text{C}_{14}\text{H}_{18}\text{O}_4$  (Found: C, 66.92; H, 7.32.  $\text{C}_{14}\text{H}_{18}\text{O}_4$  requires: C, 67.18; H, 7.25%). The IR spectrum ( $\nu_{\text{max}}^{\text{CHCl}_3}$ ,  $\text{cm}^{-1}$ : 3520, 1755 and 1710) suggested the presence of an  $\alpha,\beta$ -unsaturated  $\gamma$ -lactone containing a cyclohexanone ring and an additional alcoholic function. The  $^1\text{H NMR}$  spectrum

‡Plant material was collected in July–August on Vettore mountain, Umbria, Italy. A specimen (voucher Nos. 2115/01) has been deposited at The University of Perugia, Perugia, Italy.

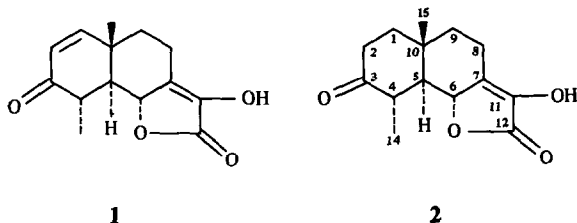


Table 1.  $^1\text{H}$  NMR spectral data of compound 2 (500 MHz,  $\text{CHCl}_3$ , TMS as internal standard)

H	$\delta$	H	$\delta$
1 ax	1.57 ddd	8 ax	2.35 dddd
1 eq	1.77 ddd	8 eq	2.89 ddd
2 eq	2.40 ddd	9 ax	1.28 ddd
2 ax	2.54 ddd	9 eq	1.71 ddd
4	2.60 ddq	14	1.22 d
5	1.27 dd	15	1.24 s
6	4.71 ddd		

$J$  (Hz): 1ax, 1eq = 14.0; 1ax, 2eq = 5.2; 1ax, 2ax = 14.0; 1eq, 2ax = 6.6; 1eq, 2eq = 2.2; 2ax, 2eq = 14.0; 4, 5 = 11.6; 4, 6 = 0.6; 4, 14 = 6.8; 5, 6 = 10.3; 6, 8ax = 1.5; 8ax, 8eq = 14.0; 8ax, 9eq = 5.4; 8ax, 9ax = 13.6; 8eq, 9eq = 1.6; 8eq, 9ax = 5.0; 9ax, 9eq = 13.6.

(90 MHz,  $\text{CDCl}_3$ , TMS as internal standard) revealed the presence of signals centred at  $\delta$ : 4.73 (1H, dd,  $J = 12$  and 0.5 Hz, collapsing to a d on irradiation at  $\delta$  2.63 and to an s(br) on irradiation at  $\delta$  1.30) consistent with the sequence  $\text{HC}-\text{CH}-\text{CH}-\text{O}-$ ; 2.63 (1H, m, partially obscured by other signals); 1.24 (3H, d,  $J = 6$  Hz collapsing to a s on irradiation at  $\delta$  2.63) accounting for  $\text{O}=\text{C}-\text{CH}(\text{CH}_3)-\text{CH}-\text{CHOR}$  and 1.24 (3H, s) for a tertiary methyl group.

All the above data showed a close similarity between compounds 1 and 2 with the lack in the latter of the  $-\text{CH}=\text{CH}-$  system.

Detailed examination of the  $^1\text{H}$  NMR spectra obtained with a 500 MHz instrument ( $\text{CDCl}_3$ ) led to the conclusion that the remaining hydroxyl group (see IR spectral data)

Table 2.  $^{13}\text{C}$  NMR chemical shifts of compound 2 (67.88 MHz,  $\text{CDCl}_3$ , TMS as internal standard)

C	$\delta$	C	$\delta$
1*	37.11 t†	8*	19.91 t
2*	19.91 t	9*	39.76 t
3	211.19 s	10	35.45 s
4	45.55 d	11	135.34 s
5	56.22 d	12	170.11 s
6	80.69 d	14*	13.60 q
7	133.63 s	15*	16.85 q

\*Assignments made by comparison with available reference compounds [1, 5].

†SFORD multiplicity.

had to be on the lactone ring as in the case of compound 1. The fact that in these spectra the signals were resolved into an approximately first-order pattern permitted selective decoupling experiments which confirmed, besides the complete assignment of every proton (see Table 1), the C-11 position of the hydroxyl group. Moreover, the particular coupling constants found for H-4, H-5 and H-6 led to the stereochemistry shown at C-4, C-5 and C-6.

The *trans*-A/B ring junction was proved by the ORD spectrum (0.002 M MeOH, 28°) which showed a positive Cotton effect ( $[\alpha]_{320} + 450^\circ$ ,  $[\alpha]_{240} + 980^\circ$ ) as reported in the literature [3, 4] for similar model compounds.

Finally, the  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ ) spectrum of 2 (see Table 2) showed 14 carbon atoms possessing complexity and chemical shifts in perfect agreement with the proposed structure.

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